Week	Lesson content
number	
1	Topic 1: Atomic Structure and The Periodic Table
2	<ul> <li>Atomic number and mass, electron configuration</li> </ul>
3	Periodicity in the Periodic Table
4	<ul> <li>Explaining trends in periods 2 and 3</li> </ul>
5	To include a WCF and MMA
6	
7	Topic 5: Formulae, Equations and Amounts of Substance
	Full and ionic equations for reactions
	Moles and the Avogadro constant
	Calculations in moles
	Calculations for titrations
	To include a WCF and MMA
8	Topic 2: Bonding  • Ionic and covalent bonding
9	
10	<ul> <li>Types of structure</li> <li>Electronegativity, bond polarity and intermolecular forces</li> </ul>
11	Formation of solutions
12	Torridation of Solutions
	To include a WCF and MMA
	Topic 3: Redox
	Oxidation and reduction
	Oxidation numbers
	Ionic half equations
	To include a WCF and MMA
13	Topic 4: The Elements of Groups 1, 2 and 7
14	Reactions of groups 1, 2 and 7
15	Explaining the trends in groups 1, 2 and 7
16	Explaining redox reactions of group 7 using oxidation numbers
17	To include a WCF and MMA
18	To include a WCF and MMA
19	Topic 6: Organic Chemistry
20	Naming of organic compounds
	Alkanes and radical substitution reactions
21	Alkenes and their reactions
22	Electrophilic addition reactions, polymer formation and their uses
23	Reactions of the haloalkanes
24	Trends in Reactivity of Nucleophilic Substitution Reactions
25	Reactions and uses of alcohols
	Carrying Out a Preparation of an Organic Liquid
	To include a WCF and MMA

Week number	Lesson content
26	Topic 8: Energetics I
27	Enthalpy changes and Hess' Law
28	. , ,
	To include a WCF and MMA
29	Topic 13: Lattice energy and Entropy
30	Born-Haber cycles, enthalpy changes
	Entropy in chemical reactions
	Gibbs free energy in chemical reactions
	To include a WCF and MMA
31	Topic 10: Equilibrium I
32	Reversible reactions and industrial processes
33	Topic 11: Equilibrium II
34	Equilibrium constants in homogenous and heterogenous systems
35	To include a MCF and MMA
	To include a WCF and MMA
	Topic 12: Acid-base equilibria
	Proton donors and acceptors
	Defining pH mathematically
	Acid dissociation constants and the ionic product of water
	Titration curves and buffer solutions
	To include a WCF and MMA
36	Topic 14: Redox II
37	Electrode potentials
38	Electrochemical cells
39	Redox equilibria
40	Redox titrations
41	To include a WCF and MMA
42	Topic 15: The Transition Elements
43	Properties of transition metals and their compounds
	Transition metal complexes and ligands
	Redox reactions of transition metals
	<ul> <li>Stability of complexes and transition metals as catalysts</li> </ul>
	To include a WCF and MMA
44	Topic 9: Kinetics I
45	Explaining how factors affect rates of reaction  Tanis 16. Kingting II.  Tanis 16. Kingting II.
46	Topic 16: Kinetics II
47	<ul><li>Rate equations</li><li>Half life and catalysts</li></ul>
48	<ul> <li>Using kinetics to investigate mechanisms.</li> </ul>
49	

Week number	Lesson content
50	To include a WCF and MMA  Topic 17: Organic Chemistry II      Stereoisomers and mechanisms     Chemistry of carbonyls     Chemistry of carboxylic acids and derivatives  To include a WCF and MMA
51 52 53 54 55	Topic 18A and B: Organic Chemistry III
56 57 58 59 60	<ul> <li>Methods to find the structure of organic compounds</li> <li>Topic 19: Modern Analytical Techniques II</li> <li>Chromatography</li> <li>NMR spectroscopy</li> <li>Identifying organic compounds using mass spectrometry</li> <li>To include a WCF and MMA</li> <li>Topic 18C: Organic Synthesis</li> <li>Planning and synethsising organic compounds</li> <li>To include a WCF and MMA</li> </ul>